
Dr. Thomas Whitcher

Prof. Andrivo Rusydi

Centre for Advanced 2D Materials, Faculty of Science, National University of Singapore, 6 Science Drive 2,
117546 Singapore

c2dwtj@nus.edu.sg

Investigation into the electronic, optical and structural properties of Perovskites

Solar cells using inorganic/organic perovskites as the primary energy-harvesting materials have recently been shown to be very efficient, and have generated ongoing investigations into developing similar materials [1-3]. Two such materials that we have investigated are the perovskites Formamidinium Lead Iodide (FAPbI₃) and Caesium Lead Bromide (CsPbBr₃) through the use of Spectroscopy Ellipsometry (SE), X-ray Absorption Spectroscopy (XAS) and first-principle calculations. The electronic and optical properties of a band-gap tunable perovskite in the form of FA_{0.85}Cs_{0.15}PbI_{2.9}Br_{0.1} show that it is more stable than the commonly used Methylammonium Lead Iodide (MAPbI₃), has a much lower exciton binding energy of (5.9 ± 2.0) meV, a strong electron-electron correlation and a much greater number of excitons and free carriers than most perovskites commonly used in solar cells, thus showing that FA_{0.85}Cs_{0.15}PbI_{2.9}Br_{0.1} is a favourable alternative to MAPbI₃ [4,5]. Most importantly, we show that the large exciton density is due to the strong electron-electron interactions within the material. The perovskite CsPbBr₃ is a promising new nano-crystalline material that has been utilised in next-generation photo-voltaic cells, emission devices and even Gamma and X-ray detection units [6-8]. The optical and electronic properties of the nano-crystalline perovskite, including the exciton binding energy, average nano-crystal size, temperature coefficients, optical transitions and electron-electron correlations are revealed. It was found that the material exists in a dual structural phase from 361K down to 7K, which is an effect of the materials nano-crystalline nature. A sudden change in the electronic structure at 150K – 160K was found to occur because the rate at which the nano-crystals transition from the tetragonal to orthorhombic phase is non-linear with temperature. This has far-reaching implications for nano-crystalline materials in nano-technology and our novel method of calculating the ratio of the structural phase duality as shown in Fig. 1 will become very useful.

References

- [1] Kojima, A., Teshima, K., Shirai, Y. & Miyasaka, T. *J. Am. Chem. Soc.* **131**, (2009) 6050
- [2] Kim, H.S. *et al. Sci. Rep.* **2**, (2012) 591
- [3] Lee, M.M. *et al. Science*, **338**, (2012) 643
- [4] Lee, J.W. *et al. Adv. Energy Mater.* **5**, (2015)
- [5] McMeekin, D.P. *et al. Science*, **351**, (2016) 151
- [6] Stoumpos, C.C. *et al. Crystal Growth & Design*, **13**, (2013) 2722
- [7] Beal, R.E. *et al. J. Phys. Chem. Lett.* **7**, (2016) 746.
- [8] Kulbak, M., Cahen, D. & Hodes, G., *J. Phys. Chem. Lett.* **6**, (2015) 2452

Figures

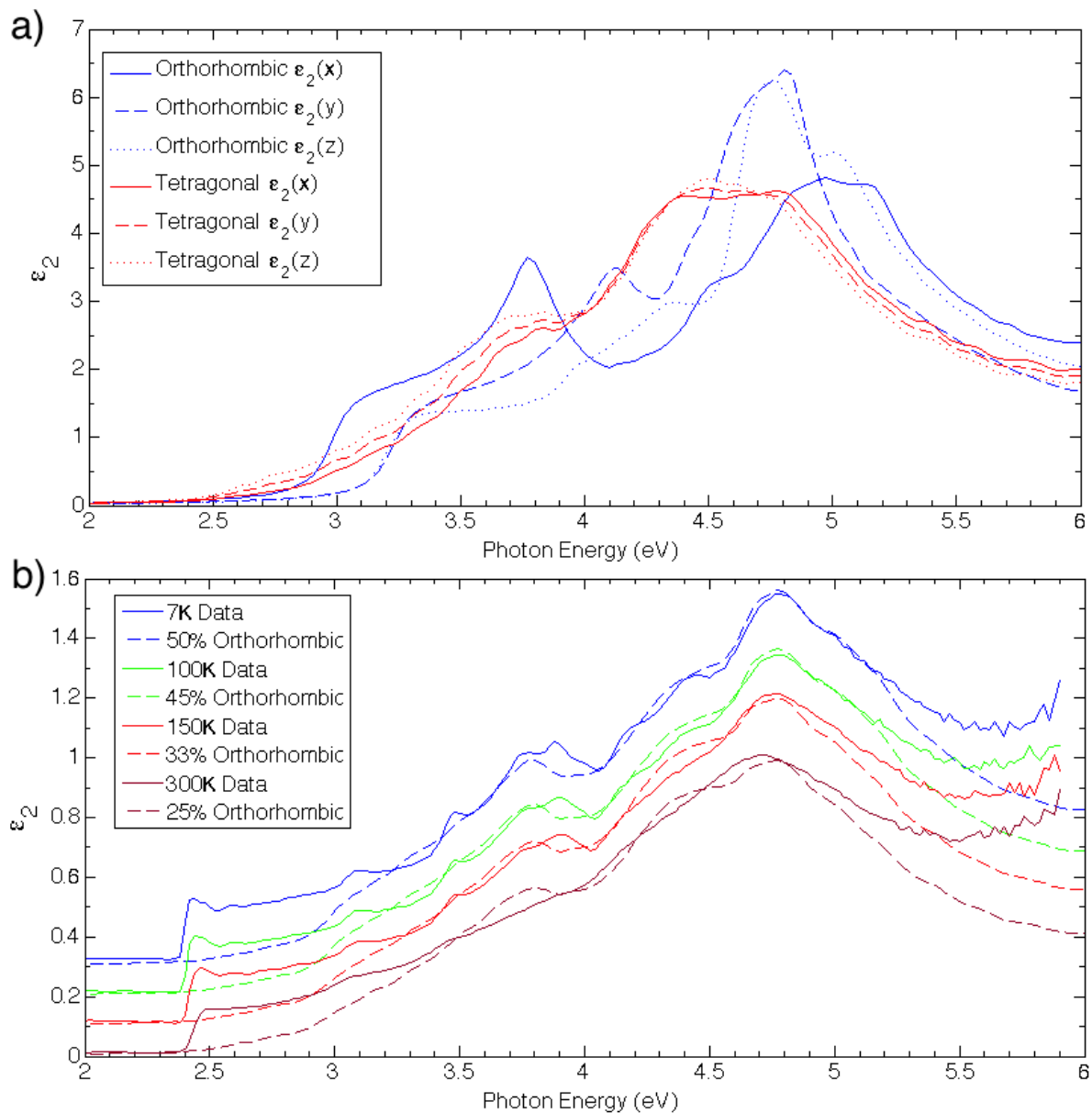


Figure 1: **a** The x, y and z components of ϵ_2 calculated from the tetragonal and the orthorhombic band structures. **b** Comparison of the combined tetragonal and orthorhombic ϵ_2 in different ratios with the data at 7K, 100K, 150K and 300K. Results are offset by 0.1 units for clarity.